A Cyclopropane Sliding Reaction of \(\Delta^3\)-Africen-10-yl Cation

NOTES

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Synopsis. Rearrangement of Δ^3 -africen-10-yl cation, generated by the treatment of 10β -chloro- Δ^3 -africene with silica gel, accompanied cyclopropane sliding.

Previously we reported^{1,2)} biogenetic like conversion of humulene (1) into $\Delta^{8(13)}$ -capnellene and dactylol. A characteristic feature of these syntheses was cyclopropane sliding reactions accompanying an apparent 1,2-shift of a methyl group. These cyclopropane sliding reactions occurred by generation of a carbocation at α -position of the cyclopropane ring. In connection with our project directed toward biogenetic like synthesis of cyclohumulanoids, we investigated the rearrangement of Δ^3 -africen-10-yl cation, generated from 10β -chloro- Δ^3 -africene (5) with an aim to establish the generality of cyclopropane sliding reaction of tricyclo[6.3.0.0^{2,4}]undecane skeleton. In this report, we wish to describe a rearrangement of 5, which resulted in cyclopropane sliding (Scheme 1).

 10β -Chloro- Δ^3 -africene (5) was synthesized from Δ^3 -

Scheme 1.

africen- 10α -ol, which had been previously prepared³⁾ from 1, via Δ^3 -africen-10 β -ol (4) by the following sequence of reactions: 1) Collin's oxidation, 2) stereoselective reduction of Δ^3 -africen-10-one (3) (90%, from 2), and 3) treatment of 3 with phosphorous pentachloride. Without any purification, unstable 5 was directly absorbed on silica-gel column. After 1 h, elution with hexane gave a mixture of three olefins (6-8). These compounds were separated by a column chromatography using silica gel impregnated with 10% AgNO₃ to give 6, 7, and 8 in 26, 1.8, and 27% yields from 3, respectively (Fig. 1). The high resolution mass spectra of these products showed the same molecular formula C₁₅H₂₂. For the purpose of elucidating its structure, 6 was converted to epoxy alcohol Oxidation of 6 with m-chloroperbenzoic acid afforded a mixture of diastereomeric diepoxides. This mixture was separated by a silica-gel column chromatography to yield major diepoxide (40%) and minor one (27%). Reduction of the minor diepoxide with lithium aluminum hydride gave rise to 9 as a single product. ¹H NMR studies on 9 including extensive decoupling experiments in the presence of a shift reagent [Eu(fod)3] revealed the existence of partial structures depicted in Fig. 2. Consideration of the possible combination of these fragments, probable reaction course and lanthanide induced shift values4) of several protons (Fig. 2) suggested that structure of the epoxy alcohol was represented by 9. Therefore, structure of the diene was figured as formula 6. ¹H NMR spectrum of 7 was so similar to that of 6 except for olefinic methyl proton (3H) in 6 and exomethylene (2H) in 7. Thus, 6 and 7 were implied to be isomers about position of double bond. From the results the molecule should be depicted as formula 7. Furthermore, examination of ¹H NMR spectrum of 8 and the corresponding diepoxide 10, which was from **8** by epoxidation with

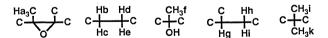


Fig. 2. The following partial structure of **9** was revealed by ¹H NMR spectra in the presence of Eu(fod)₃ (Eu/**9**=0.37).

Lanthanide induced shift values $[\Delta\delta/([Eu]/[9])]$ a=4.8, b=9.7, c=10.4, d=9.7, e=15.4, f=5.0, g=31.1, h=8.3, i=7.2, j=1.6, k=3.9.

$$C \stackrel{\mathsf{CH}_3}{+} C \stackrel{\mathsf{H}_3}{\overset{\mathsf{C}}{\overset{\mathsf{C}}{\overset{\mathsf{H}_3}}}} \stackrel{\mathsf{H}_3}{\overset{\mathsf{C}}{\overset{\mathsf{C}}{\overset{\mathsf{H}_3}}}} \times 2 \qquad \stackrel{\mathsf{H}}{\overset{\mathsf{H}}{\overset{\mathsf{H}_3}{\overset{\mathsf{H}_3}}}} \stackrel{\mathsf{H}}{\overset{\mathsf{H}_3}} \stackrel{\mathsf{H}}{\overset{\mathsf{H}_3}} \stackrel{\mathsf{H}_3}{\overset{\mathsf{H}_3}} \stackrel{\mathsf{H}_3}} \stackrel{\mathsf{H}_3}{\overset{\mathsf{H}_3}} \stackrel{\mathsf{H}_3}{\overset{\mathsf{H}_3}} \stackrel{\mathsf{H}_3}{\overset{\mathsf{H}_3}} \stackrel{\mathsf{H}_3}{\overset{\mathsf{H}_3}} \stackrel{\mathsf{H}_3}{\overset{\mathsf{H}_3}} \stackrel{\mathsf{H}_3}} \stackrel{\mathsf{H}_3} \stackrel{\mathsf{H}_3} \stackrel{\mathsf{H}_3}} \stackrel{\mathsf{H}_3} \stackrel{\mathsf{H}_3} \stackrel{\mathsf{H}_3}} \stackrel{\mathsf{H}_3} \stackrel{\mathsf{H}_3} \stackrel{\mathsf{H}_3} \stackrel{\mathsf{H}_3}} \stackrel{\mathsf{H}_3} \stackrel{\mathsf{H}_3}} \stackrel{\mathsf{H}_3} \stackrel{\mathsf{H}_3}} \stackrel{\mathsf{H}_3} \stackrel{\mathsf{H}_3}} \stackrel{\mathsf{H}_3} \stackrel{\mathsf{H}_3}} \stackrel{\mathsf{H}_3} \stackrel{\mathsf{H}_3}} \stackrel{\mathsf{H}_3} \stackrel{\mathsf{H}_3$$

Fig. 3. Partial structures in the molecule of 8.

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chloroperbenzoic acid (72%) as a sole product, revealed that the absence of cyclopropane and the presence of partial structures depicted in Fig. 3 in the molecule of **8**. Considering the above results and possible reaction process, the structure of the triene was presumed to be expressed by the formula **8**.

As described above, it was found that cyclopropane sliding was also took place in acidic rearrangement of 10β -chloro- Δ^3 -africene (5). These results combined with the previous one could suggest that carbocations generated at C-1 or C-5 position of tricyclo-[6.3.0.0^{2,4}]undecane skeleton (α -carbon of the cyclopropane ring) rearranged through cyclopropane sliding in general.

Experimental

¹H NMR and ¹³C NMR spectra were recorded on a HITA-CHI R20B (60 MHz) and a JEOL JNM FX-100 (25 MHz) instrument using TMS as an internal standard, respectively. High-resolution mass spectra were measured on a JEOL JMS D-300 mass spectrometer. All reactions were carried out under argon atmosphere and stirring.

t-2,6,6,9-Tetramethyl-trans-1-transoid-1,2-cis-2-tri- $\operatorname{cyclo}[6.3.0.0^{2,4}]$ undec-9-en-r-5-ol (4). (10-Hydroxyafricene). To a suspension of Collin's reagent (630 mg, 2.4 mmol) and dry Celite (1.0 g) in dry CH₂Cl₂ (10 ml) was added a solution of 2 (150 mg, 0.55 mmol) in CH₂Cl₂ (2 ml) at room temperature. After stirring for 0.5 h at room temperature, the reaction mixture was diluted with Et2O (20 ml) and filtered through Celite bed. After the filtrate was concentrated in vacuo, the crude oil was dissolved in THF (5.0 ml) and LiAlH₄ (20 mg, 0.53 mmol) was added to the solution at room temperature. After stirring at room temperature for 1 h, the solution was cooled to 0°C, diluted with Et₂O, quenched by the addition of ice, and dried over Na₂SO₄. After filtration, the solvent was removed in vacuo and the residue was chromatographed on silica-gel column (4% EtOAc-C₆H₆) to yield a pure product 4 (135 mg, 90%): ¹H NMR (CDCl₃) δ =1.01 (3H, s), 1.05 (6H, s), 1.60 (3H, bs), 3.84 (1H, d, J=6 Hz), 5.24 (1H, bs); IR (neat) 3600—3200 cm⁻¹. Found: m/z 220.3537. Calcd for C₁₅H₂₄O: M, 220.3540.

rel-(3R,5S,8R)-2,6,6,9-Tetramethyltricyclo[6.3.0.0^{3,5}]undeca-1,9-diene (6), 6,6,9-Trimethyl-2-methylene-trans-1-transoid-1,3-cis-3-tricyclo[6.3.0.0^{3,5}]undec-9-ene (7), and trans-2,6,6,9-Tetramethylbicyclo[6.3.0]undeca-2,4,9-triene (8). To a solution of PCl_5 (162 mg, 0.74 mmol) in a mixture of C_6H_6 (27 ml) and $C_6H_5CH_3$ (27 ml) was added 185 mg (0.89 mmol) of 4 at 0 °C and the mixture was stirred for 1 h at room temperature. The reaction mixture was quenched with saturated NaHCO₃ aqueous solution (25 ml), and extracted with C_6H_6 three times. The combined extracts were washed with water, dried over Na_2SO_4 and concentrated in vacuo to give a pasty mass which was absorbed on SiO_2 (WAKO silica gel, 8 g) with hexane. After 1 h, mixture of hydrocarbons was eluted with hexane. Separation of the mixture with $AgNO_3$ silica-gel column

(0.5% EtOAc-hexane) afforded **6** (44 mg, 26%), **7** (3 mg, 1.8%), and **8** (45 mg, 27%). **6**: ¹H NMR (CDCl₃) δ =0.6—1.1 (4H, m), 1.0, 1.2 (each 3H, s), 1.65 (6H, bs), 5.29 (1H, bs). Found: m/z 202.3386. Calcd for $C_{15}H_{22}$: M, 202.3388. **7**: ¹H NMR (CDCl₃) δ =0.2—0.8 (3H, m), 0.93, 1.23 (each 3H, s), 1.63 (3H, bs), 4.73 (2H, bs), 5.29 (1H, m). Found: m/z 202.3384. Calcd for $C_{15}H_{22}$: M, 202.3388. **8**: ¹H NMR (CDCl₃) δ =1.08, 1.13 (each 3H, s), 1.65, 1.75 (each 3H, bs), 5.2, 5.9 (4H, m); ¹³C NMR (CDCl₃) δ =14.8 (q), 19.7 (q), 23.6 (q), 33.4 (s and t, 2C), 34.8 (q), 40.8 (t), 45.4 (d), 46.8 (d), 123.0 (d), 123.3 (d), 125.4 (d), 139.5 (s), 140.1 (d), 141.3 (s). Found: m/z 202.3382. Calcd for $C_{15}H_{22}$: M, 202.3388.

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rel-(1S,2R,3R,5S,8R,9S)-1,2-Epoxy-2,6,6,9-tetramethyltricyclo[6.3.0.0 3,5]undecan-9-ol (9). To a solution of mchloroperbenzoic acid (183 mg, 1.01 mmol) in CH2Cl2 (20 ml) at 0°C was added 97 mg (0.48 mmol) of 6. After stirring for 1 h at room temperature, usual workup followed by chromatographic separation with silica-gel column (2% EtOAc-hexane) afforded minor diepoxide (30 mg, 27.7%): ¹H NMR (CDCl₃) δ =0.0—0.9 (3H, m), 0.99, 1.11, 1.3, 1.4 (each 3H, s), 1.74 (1H, d, J=15 Hz), 2.22 (1H, dd, J=15 and 1.5 Hz), 3.3 (1H, bs), and major diepoxide (45 mg, 40%): ¹H NMR (CDCl₃) δ =0.1—1.0 (4H, m), 1.0, 1.1, 1.33, 1.4 (each 3H, s), 1.8 (1H, dd, I=16 and 2.5 Hz), 2.33 (1H, d, J=16 Hz), 3.3 (1H, d, J=2.5 Hz). A solution of the minor diepoxide (30 mg, 0.13 mmol) in THF (0.5 ml) was added to a well stirred slurry of LiAlH₄ (5.4 mg, 0.14 mmol) in THF (15 ml) at 0°C. After stirring for 2 h at room temperature, the excess of reagent was decomposed by the addition of icewater and the mixture was extracted with EtOAc three times. The combined extracts were dried over Na₂SO₄ and evaporated in vacuo to give paste, which was purified with chromatography on a silica-gel column (10% EtOAc-C₆H₆) to afford 9 (27 mg, 89%): IR (neat) 3600—3200 cm⁻¹; ¹H NMR $(CDCl_3) \delta = 0.0 - 0.9 (3H, m), 0.99, 1.11, 1.18, 1.35 (each 3H, m)$ Found: m/z 236.3531. Calcd for $C_{15}H_{24}O_2$: M, 236.3534.

rel-(1R,2S,3R,8S)-2,3,9,10-Diepoxy-2,6,6,9-tetramethylbicyclo[6.3.0]undec-4-ene (10). To a solution of *m*-chloroperbenzoic acid (56 mg, 0.33 mmol) in CH₂Cl₂ (12 ml) at 0 °C was added **8** (30 mg, 0.15 mmol) and the reaction mixture was stirred for 2 h at room temperature. Usual workup followed by chromatographic purification with silica-gel column (2% EtOAc-hexane) gave a pure diepoxide **10** (25 mg, 72%): ¹H NMR (CDCl₃) δ=1.1, 11.2, 1.3, 1.35 (each 3H, s), 3.26 (1H, bs), 3.3 (1H, s), 5.39 (1H, d, J=11 Hz), 5.51 (1H, d, J=11 Hz). Found: m/z 234.3374. Calcd for C₁₅H₂₂O₂: M, 234.3376.

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